

Molecular Geometry Study Guide

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CHAPTER 7 – CHEMICAL BONDING AND MOLECULAR GEOMETRY

Web7. DEFINE THE MOLECULAR ORBITAL (MO) MODEL FOR BONDING. 8. RECALL THE ENERGY LEVEL DIAGRAM PATTERNS (TWO TYPES) AND HOW TO LABEL THE MOLECULAR ORBITALS FOR DIATOMIC HOMONUCLEAR AND HETERONUCLEAR SPECIES (DIATOMIC IONS OR MOLECULES). 9. WRITE THE GROUND STATE ELECTRON CONFIGURATION LISTING ALL BONDING AND ANTI-BONDING MOLECULAR ORBITALS. 10.

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Web2. THE ELECTRON GEOMETRY WILL BE THE SAME OVERALL SHAPE. (THE LONE PAIR IS AN ELECTRON GROUP, JUST LIKE THE BONDS, AND IT WILL OCCUPY ONE OF THE THREE SPACES). MOLECULAR GEOMETRY LOOKS AT THE GEOMETRY OF THE BONDS ONLY AND IGNORES THE LONE PAIR. 3. THE ANGLES IMMEDIATELY NEXT TO THE "HIGH-DENSITY" GROUP WILL BE GREATER THAN THEY WERE. THE

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WebSECTION 2 COVALENT BONDING AND MOLECULAR COMPOUNDS OBJECTIVES 1. DEFINE MOLECULE AND MOLECULAR FORMULA. 2. EXPLAIN THE RELATIONSHIPS AMONG POTENTIAL ENERGY, DISTANCE BETWEEN APPROACHING ATOMS, BONDS LENGTH AND BOND ENERGY. 3. STATE THE OCTET RULE. 4. LIST THE SIX BASIC STEPS USED IN WRITING LEWIS DOT STRUCTURES. 5.

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